

## Hydrogen Production & Distribution

- **PRODUCTION & COSTS**– Hydrogen ( $H_2$ ) is an energy carrier. It can be produced from all forms of energy and used for power generation or as a transport fuel, mainly in association with fuel cells. Natural gas and coal are currently the cheapest sources of  $H_2$  and will remain so. Because production processes release  $CO_2$ , however,  $CO_2$  capture and storage (CCS, ETE01) are vital to reduce emissions.  $H_2$  production from renewable and nuclear energy will need more R&D and more time to enter the market. Decentralised  $H_2$  production is the best choice for market uptake and for avoiding costly distribution infrastructure. But it is less efficient and costs more than large-scale, centralised production. The cost of decentralised  $H_2$  production may exceed US \$50/GJ $H_2$ <sup>1</sup> today. Over the coming decades, improved technologies could provide fossil fuels-based  $H_2$  at \$10-15/GJ, including the cost of CCS.
- **DISTRIBUTION** – Because of the low volumetric energy density of  $H_2$ , its distribution for energy use is rather expensive and energy-intensive. Investment and pumping-power requirements are greater than for natural gas. Large-scale  $H_2$  distribution by pipeline adds \$1-\$2/GJ to  $H_2$  production costs. Distribution of liquid  $H_2$  is more costly (\$7-\$10/GJ) as energy is needed for liquefaction at  $-253^\circ C$ . Refueling stations may add \$3-\$9/GJ to  $H_2$  costs.
- **STORAGE IN FUEL CELL VEHICLES (FCV)** – On board  $H_2$  storage presents R&D challenges (see ETE06, *Fuel Cells*). The compactness, driving range and cost of current options do not yet meet the requirements. Gaseous storage (350-700 bar) and liquid storage ( $-253^\circ C$ ) are commercially available, but energy-consuming and costly. Electricity needed for gas and liquid storage represents, respectively, some 12% and 35% of the  $H_2$  energy content. The tank costs \$3,000-\$4,000 per vehicle. Solid storage promises potential breakthroughs, but much R&D is needed.
- **INFRASTRUCTURE** – Global investment to supply  $H_2$  to the world's transport sector could be in the range of several hundred billion dollars over several decades (\$0.1-\$1.0 trillion for pipelines and \$0.2-\$0.7 trillion for refueling stations). This level of investment is not insurmountable in the long term, but building infrastructure today is premature because key  $H_2$  technologies that may have an impact on infrastructure are still under development.
- **POTENTIAL & BARRIERS**–  $H_2$  could gain significant market share in the transport sector if costs of production, distribution and end-use technologies (e.g., fuel cells) decrease according to expectations, and if strong policies are put in place to reduce  $CO_2$  emissions. Under favorable assumptions,  $H_2$  could be entering the market around 2020 and powering some 700 million fuel cell vehicles by 2050 (30% of projected global fleet). Besides costs, other barriers to  $H_2$  use for energy applications include dedicated infrastructure needs, as well as competition with other emerging technologies and fuel options such as biofuels and battery-electric vehicles. However, since no single fuel or technology is likely to meet fast growing demand for clean transport fuels, various options are expected to play complementary roles in diversified regional markets.

**PROCESSES & COSTS** - Hydrogen ( $H_2$ ) is an energy carrier that can be obtained from different sources: fossil fuels (natural gas reforming, coal gasification); renewable and nuclear energy (biomass processes, photo-electrolysis, biological production, high-temperature water splitting); and electricity (water electrolysis). If  $H_2$  is produced from renewable and nuclear energy, or from natural gas and coal with  $CO_2$  capture and storage (CCS), then it can be largely carbon-free, thus helping to reduce  $CO_2$  emissions and diversify the energy supply. If  $H_2$  is produced by water electrolysis, emissions are created by associated upstream electricity generation. At present,  $H_2$  is produced largely from fossil fuels without CCS (48% from natural gas, 30% from refinery/chemical off-gases, 18% from coal, the rest from electrolysis). Most of today's production (some 65 million tonnes per year) is for captive use in the chemical and refinery industries. In the future,  $H_2$  could be used for power generation and in transport by fueling gas turbines, fuel cells and combustion engines. Used in FCV,  $H_2$  could significantly increase efficiency and emission reduction in transport. However, using  $H_2$  for energy applications requires more

efficient, less costly production processes, ideally with no  $CO_2$  emissions. Decentralised production is the best choice for market uptake as it minimises the needs for distribution infrastructure. But it is less efficient than large-scale, centralised production, and it makes CCS impractical.



Some 140  $H_2$  refueling stations world wide fuel 400 FCV and 100 buses for demonstration projects

■ **Electrolysis** is a well-known electro-chemical process to split water into H<sub>2</sub> and oxygen (O<sub>2</sub>) using electricity. Alkaline electrolyzers with potassium hydroxide (KOH) electrolyte are commercially available. Efficiency is a key parameter for electrolysis, as costs are largely determined by electricity costs. Best-practice efficiency could be higher than 85% (GJH<sub>2</sub>/GJ<sub>el</sub>), but commercial devices achieve between 55% and 75%. New advanced electrolyzers may approach the upper limit. At high temperatures, heat consumption increases while electricity needs decrease. High-temperature electrolysis (800°C-1,000°C) may therefore offer higher efficiency, in particular using residual heat. Also, high-pressure electrolysis can make H<sub>2</sub> pressurisation unnecessary and improve efficiency. New electrolyser concepts are based on fuel cells working in reverse mode. Small-scale polymer electrolyte membrane FC (PEMFC) electrolyzers (60°C-80°C, 15 bar, 50% efficiency) are commercially available. Solid oxide FC (SOFC) electrolyzers functioning at 700°C-1,000°C need more research. Current electrolysis costs are typically above \$30/GJH<sub>2</sub>, but could drop to below \$20/GJ (including pressurisation) over coming decades, assuming electricity at \$35/MWh and 80% process efficiency. Use of off-peak electricity and large-scale plants may reduce costs, although the cost of CCS is expected to increase the cost of electricity.

■ **Natural gas reforming** is a mature technology used in the refinery and chemical industries for large-scale H<sub>2</sub> production. Small-scale reformers are currently used in demonstration H<sub>2</sub> refuelling stations (decentralised production). Reforming options include catalytic steam methane reforming (SMR), partial oxidation (PO) and other variants under development. In SMR, methane reacts with steam at 700°C-850°C to produce syngas, a mixed H<sub>2</sub> and carbon monoxide (CO) gas. CO is then converted into CO<sub>2</sub>, producing additional H<sub>2</sub> by water-gas shift reaction. In the PO process, methane reacts initially with pure O<sub>2</sub> to provide syngas. SMR offers efficiencies of up to 80%-85% in large-scale units (excluding H<sub>2</sub> compression). If residual steam is re-used, total efficiency may be higher. Small units have lower efficiency (at least 10-15 percentage points lower) and higher unit costs. Producers have recently much improved the compactness of small-scale reformers (10x3x3m) and their capacity (5.5-7.5 GJ/hour), but further R&D is needed to reduce costs and increase efficiency. H<sub>2</sub> compression and CCS (eventually, in large units) may each further reduce net efficiency by some 5-10 percentage points. CCS in small plants is probably not practical. At current natural gas prices, (\$6-\$9/GJ), the cost of H<sub>2</sub> from natural gas reforming ranges from between \$10 and \$15/GJH<sub>2</sub> (in large-scale production for captive use) to more than \$30/GJ, with high sensitivity to natural gas prices, processes and economy of scale. Small-scale decentralised production may exceed \$50/GJ. Compressed H<sub>2</sub> in tubes may cost \$90-\$100/GJ (delivered). Assuming natural gas at \$3-\$4/GJ, cost projections for natural gas reforming range from \$6 in

large plants to \$10-\$12/GJ H<sub>2</sub> in small plants (*Prospects for Hydrogen and Fuel Cells*, IEA 2005). Projected CCS costs are expected to add \$1-\$3/GJ, depending on process and scale.

■ **Coal gasification** produces a gas mixture of H<sub>2</sub>, CO, carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>). The CO can then be converted into CO<sub>2</sub> and additional H<sub>2</sub> through a water-gas shift reaction. In addition to H<sub>2</sub>, the final product offers relatively pure CO<sub>2</sub>, ready for pressurisation and storage (CCS). Final H<sub>2</sub> purification is needed for most applications. Although a mature process, coal gasification is currently more expensive than natural gas reforming because of the gasifier and the need for O<sub>2</sub> for the reaction process. Large-scale, integrated gasification combined cycles (IGCC) are considered an attractive option for centralised co-generation of electricity and H<sub>2</sub>, with comparably low CCS costs. Assuming costs of \$1-\$1.5/GJ for coal and \$35-\$40/MWh for electricity, with 45% electrical efficiency, projected H<sub>2</sub> production cost with CCS would range between \$7 and \$10/GJH<sub>2</sub> (IEA 2005). IGCC demonstration plants are operating today in several countries to produce electricity (no H<sub>2</sub>). They have proved more expensive and less reliable than conventional coal power plants. Cheaper gasifiers and new processes to produce O<sub>2</sub> could make IGCC plants more economically attractive. H<sub>2</sub> is also produced as a by-product from catalytic reforming in refineries, or through off-gas reforming in petrochemical plants, also from ethylene crackers, from chlorine plants and from coke oven gas.

■ **Thermal water-splitting** occurs at very high temperatures exceeding 2,500°C, but thermochemical processes such as sulphur-iodine (S-I) or bromine-calcium (Br-Ca) cycles may reduce temperatures to below 1,000°C. These processes require low-cost high-temperature heat from nuclear or solar sources, also corrosion-resistant materials. The S-I process is the most promising, with about 43% LHV efficiency and an operating temperature of 950°C. Cost projections suggest a H<sub>2</sub> cost of \$10-\$20/GJ using nuclear heat from nuclear high-temperature gas reactors (HTGR), and a cost of \$20-\$30/GJ using heat from advanced, megawatt-scale concentrating solar power (CSP) systems. Both technologies are unlikely to be commercial before 2030.

■ **H<sub>2</sub> from Biomass** is the only direct way to produce H<sub>2</sub> from renewable energy without major technology breakthroughs. Biomass can be converted into H<sub>2</sub> via various processes (pyrolysis, gasification, anaerobic digestion etc.). While R&D focuses on gasification, synergies with other fuel production processes (biofuels) could open the way to other options and accelerate market uptake. But H<sub>2</sub> production from biomass would compete with biofuels and CHP production. In general, as basic feedstock availability is limited, production from biomass will not benefit from large economies of scale. Costs are expected to be high compared with coal gasification or gas reforming.

■ **Photo-electrolysis** produces H<sub>2</sub> using

sunlight to illuminate a water-immersed semiconductor that converts the light into chemical energy to split water into H<sub>2</sub> and O<sub>2</sub>. This method promises lower capital costs than combined photovoltaic-electrolysis systems and it holds considerable potential for technology breakthroughs. Test-scale devices have shown solar-to-H<sub>2</sub> conversion efficiencies of up to 16% (IEA, 2005). But cost estimates are premature. ■ **Biological processes** derive H<sub>2</sub> from organic matter using micro-algal photo-synthesis and cyanobacteria. These processes require genetic engineering to achieve significant levels of H<sub>2</sub> production. Much research is still needed to demonstrate feasibility. ■ **Projected H<sub>2</sub> production costs** in Table 1 reflect a range of different technologies, economies of scale and energy prices.

**DISTRIBUTION** – ■ **Pipelines** are considered the only option to move large amounts of H<sub>2</sub>. They have been used to transport H<sub>2</sub> for more than 70 years. Several thousand kms of H<sub>2</sub> pipelines are currently in operation world wide. The energy required to pump H<sub>2</sub> through pipelines is some 4.5 times higher than for natural gas per unit of delivered energy. As a consequence, long distances H<sub>2</sub> transportation for energy use may not be economically competitive. Transportation costs to deliver gaseous H<sub>2</sub> to refueling stations are in the range of \$1-\$2/GJ, assuming that H<sub>2</sub> compression to refueling pressure is included in the cost of the refueling station. ■ **Liquid H<sub>2</sub>** transport by truck, rail or ship is more expensive than gas piping. In current plants, the electricity required for H<sub>2</sub> liquefaction at -253°C is about 35-43 MJ<sub>el</sub>/kgH<sub>2</sub>, with potential for future reduction to 25 MJ<sub>el</sub>/kg. The cost of liquefaction in large systems is about \$7-\$9/GJ, 75% of which comes from the cost of electricity. Transportation of liquid H<sub>2</sub> by ship over long distances is also more expensive than for natural gas (LNG) since very low-temperature cryogenic technology is needed. Fast ships are required to reduce boil-off losses (0.2%-0.4% of liquid H<sub>2</sub> per day, which could be recovered, however, and used to fuel the ship). ■ **Refueling Stations** – Some 140 H<sub>2</sub> refueling stations are in operation world wide (90 under construction) to fuel some 400 FCV and 100 buses used in demonstration projects. Most stations deliver gaseous H<sub>2</sub> at 350 bar. H<sub>2</sub> is either produced on-site from electrolysis or steam reforming, or received from centralised plants. Costs of refueling stations are estimated between \$3/GJH<sub>2</sub> and \$10/GJ with centralised H<sub>2</sub> production and on-site production, respectively. These costs include investment and H<sub>2</sub> compression. Transportation, distribution and refueling stations may add some \$5-\$12/GJ to H<sub>2</sub> production costs.

**STORAGE IN FCV** – On-board H<sub>2</sub> storage for fuel cell vehicles (FCV) is challenging and may have significant impact on H<sub>2</sub> distribution infrastructure and standards (e.g. operating pressure). The target is to store 4-5 kg of H<sub>2</sub> (sufficient for a drive range of 400-500 km) while minimising volume, weight, storage energy, cost, and

refueling time, and providing prompt H<sub>2</sub> release on demand. Storage requires energy-intensive compression at high pressure (350-700 bar) or liquefaction at -253°C. Electrical energy required for compression or liquefaction represents, respectively, some 12% or 30% of the H<sub>2</sub> energy content (LHV). Current commercial options do not fully meet requirements for compactness, drive-range, and cost. Liquid or gaseous storage at 700 bar both require more space than gasoline with equivalent energy content. The tank costs more than \$3,000-4,000 per vehicle. H<sub>2</sub> storage in solid materials may offer decisive advantages, but this is still under development, with a number of materials under investigation. On-board reforming to produce H<sub>2</sub> from fossil fuels has also proved challenging and expensive.

**INFRASTRUCTURE** - Estimates of H<sub>2</sub> infrastructure investment are complicated by significant uncertainty. The cost of H<sub>2</sub> supply infrastructure for road transport is estimated to be in the order of several hundred billion dollars. Assuming large-scale, centralised H<sub>2</sub> production, the cost of worldwide pipeline-based distribution systems for road transport could range from \$0.1 to \$1.0 trillion. The *incremental* investment in refueling stations would be somewhere between \$0.2 for centralised H<sub>2</sub> production and \$0.7 trillion for decentralised production. A full H<sub>2</sub> economy (i.e., widespread use of H<sub>2</sub> in transport and stationary sectors) would require global pipeline investment in the order of \$ 2.5 trillion, the bulk of which would be to finance supplying commercial and residential customers. Assuming early retirement or partial replacement of existing natural gas pipelines, a significant part of this cost would be *incremental*. The level of investment needed for H<sub>2</sub> infrastructure is not insurmountable when compared with the \$20-trillion investment in energy supply systems that is estimated to be needed if growth in energy demand up till 2030 is to be met (IEA *World Energy Outlook* 2006).

**POTENTIAL & BARRIERS** - According to *Prospects for Hydrogen and Fuel Cells* (IEA, Dec. 2005) and *Energy Technology Perspectives* (IEA, June 2006), H<sub>2</sub> is likely to gain significant market share over the coming decades if the cost of H<sub>2</sub> production, distribution and end-use fall significantly, and if effective policies are put in place to increase energy efficiency, mitigate CO<sub>2</sub> emissions and improve energy security. H<sub>2</sub> production costs should be reduced by a factor of 3 to 10 (depending on technologies and processes) and fuel cell cost by a factor of 10 or more. At the same time, emission reduction incentives of \$25-\$50/tCO<sub>2</sub> (depending on fossil fuel price) would help to make H<sub>2</sub>, fuel cells and other clean energy options more competitive economically. Under these assumptions, emissions growth over the coming decades could be reduced in proportions that would bring annual emissions in 2050 down to half those projected in a business-as-usual scenario. Use of H<sub>2</sub> for energy applications would grow during the years starting from 2020 to reach some 12.5 EJ per year (0.3 Gtoe) by 2050, concentrated mostly on

the transport sector. Thanks to the high efficiency of FCVs, this relatively limited input of H<sub>2</sub> (2%-3% of projected total primary energy supply) could fuel some 30% of the global fleet of passenger cars (about 700 million cars). If H<sub>2</sub> for FCV is combined with H<sub>2</sub> used in other applications (refinery and chemicals industries), total H<sub>2</sub> use by 2050 would amount to some 22 EJ (almost four times today's annual use of H<sub>2</sub>). Under less optimistic assumptions regarding technology and CO<sub>2</sub> reduction policies, H<sub>2</sub> is unlikely to gain significant

market share as alternative fuel and technology options (biofuels, Fischer-Tropsch synfuels, hybrids, battery-electric vehicles, etc.) could play a more important role in future. In addition to costs and competition from other technologies, barriers to H<sub>2</sub> market uptake include the need for dedicated infrastructure. However, no single fuel or technology is likely to meet the expected fast-growing demand for clean transport fuels. Various options are therefore expected to play complementary roles in regionally diversified markets.

**Table 1 - Typical Data and Figures for H<sub>2</sub> Production & Distribution Technologies**

Data Confidence – Industrial H <sub>2</sub> production is based on well known technologies, but new processes with higher efficiency, lower costs and eventually CCS are needed to produce H <sub>2</sub> for energy use. Typical figures for these technologies are more uncertain. H <sub>2</sub> costs are highly sensitive to coal, gas, biomass and electricity prices.								
Current H <sub>2</sub> Annual Production: 65 million tonnes per year, equivalent to 8EJ (less than 2% of world total primary energy supply); 48% from natural gas, 30% refinery-gas/chemicals, 18% coal, 4% electrolysis								
Efficiency of H <sub>2</sub> production from electrolysis (incl. auxiliaries, no compression)								
Technology	Alkaline large-scale	Alkaline high-pressure	Advanced Alkaline	PEM	SOFC			
Status	Commercial	Commercial	Precommercial	Precommercial	Prototype			
T (°C)	70-90	70-90	80-140	80-150	900-1000			
P (bar)	atm. to 25	up to 690	up to 120	up to 400	up to 30			
kWh/kgH <sub>2</sub>	48-60	56-60	42-48	40-60	28-39			
Efficiency of H <sub>2</sub> production from natural gas reforming (figures in brackets do not include compression)								
Technology	Steam reforming, large-scale 50 PJH <sub>2</sub> /yr, 80 bar		Steam reforming small-scale 0.02 PJH <sub>2</sub> /yr, 340 bar		Partial Oxidation	Auto-thermal Reforming		
Status	Commercial no CCS	Future with CCS	Commercial no CCS	Future no CCS	Commerc. no CCS	Precommenc. no CCS		
Effic. %, LHV	72-77 (76-80)	61-70 (62-78)	47-55 (60-65)	60-65 (70-75)	66-76	66-73		
Efficiency of H <sub>2</sub> production from coal gasification w/out electricity cogeneration (IGCC) - H <sub>2</sub> at 75 bar								
Technology	Current Gasification		O <sub>2</sub> -blown Gasification	Adv.CO <sub>2</sub> Membr.Sep	Cogeneration		Cogeneration Membr. Sep.	
Effic. %, LHV	no CCS	CCS	no CCS	CCS	no CCS	CCS	CCS	
	57	51	67	62	64	83	70	77
Current and Projected H <sub>2</sub> Production Costs								
<b>Current H<sub>2</sub> production cost (US\$/GJ) Sensitivity to energy price</b> 				<b>Projected H<sub>2</sub> production cost 2020-2030 (US\$/GJ) Sensitivity to technologies &amp; processes</b> 				
<b>Further Information</b> - <a href="http://www.iea.org">www.iea.org</a> ; <a href="http://www.iea-hia.org">www.iea-hia.org</a> ; <a href="http://www.hfpeurope.org">www.hfpeurope.org</a> ; <a href="http://www.iphe.net">www.iphe.net</a> ; Prospects for Hydrogen and Fuel Cells (IEA, 2005); Energy Technology Perspectives (IEA, 2006); Hydrogen Economy: Opportunities, Costs, Barriers and RD&D Needs, NRC (2004).								